

BIOGEOCHEMISTRY OF NUTRIENTS, TRACE METALS AND ORGANIC MICROPOLLUTANTS IN THE NORTH SEA.

SUBPROJECT: BIOGEOCHEMISTRY OF NUTRIENTS AND TRACE METALS IN THE NORTH SEA.

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Objectives:

>> our research : biogeochemical cycling in aquatic systems

- T North Sea
- T Scheldt estuary
- **T** Scheldt and tributaries
- T marine biota in the North Sea
- quantification and qualification of the nitrogen cycle
- study of dissolved trace metals in the water column
- special attention for biogeochemical cycling of Hg and As

North Sea:



spatio-temporal trends in nutrient concentrations





◆ <u>nitrate</u> concentration: shows high variation

- average : $18.3 \,\mu M$ range: DL $139 \,\mu M$
- <u>phosphate</u> concentration :
 - rather low throughout the year
 - average: 0.8 µM
- <u>Silicate</u> concentration:
- average: $8 \mu M$ range: DL $32 \mu M$
 - ◆ <u>temporal behavior:</u> -example: nitrate
 - clear seasonal trend : highest [nitrate] in winter, lowest in spring and summer
 - ▶ reconstitution of winter stock
 - ▶ biological activity



North Sea:

spatio-temporal trends in nutrient concentrations

Example : dissolved silicate



Low salinity zone is strongly influenced by the Scheldt outflow and coastal inputs

significantly higher nutrient poolscompared to the high salinity zone

Channel and central North Sea: low nutrient pools

Clear seasonal and spatial variation in nutrient concentrations in the North Sea

Nitrogen pools in the North Sea





- ♦ The ammonium pool is almost negligible
 - Most of the time: the dissolved organic nitrogen pool (DON) is dominant
 - The DON pool can account for up to 95% of the N pool

>> The DON pool in the North Sea plays an important role in the nitrogen cycle

Absolute uptake rates





- selected station: 330
 - Nitrate uptake rates are higher than those for ammonium and urea
 - highest nitrate uptake in spring
 highest ammonium uptake at the end of summer
 - Iow ammonium uptake rates
 - Iowest urea uptake rates

The Scheldt estuary:

The Scheldt receives high amounts of waste (mostly untreated)

>> causes high microbial activity and anoxic conditions upstream



gradients of O₂, pH and ionic strength influence mass transport to the sea



Nutrient distribution in the Scheldt estuary





Trend in NH₄ significantly related with oxygen depletion

- nutrient concentrations are very high
 - decreasing concentrations with increasing salinity
 - ♦ longitudinal profile has a linear shape

Biogeochemical behavior of redox sensitive trace elements

- Example : dissolved As-transport
- longitudinal profile has a convex shape
- **b** suggests strong mobilization at freshwater/ salt water interface
- **indicates influence of biological activity**
- **b** evidence was also found for influence of Fe and Mn redox chemistry







Marine Biota :

- 25 species of seafish and shellfish were investigated for total Hg and As
- BUT : Hg and As occur in a wide range of species, several of them more toxic than others THUS : speciation needed !
- ▶ speciation techniques were developed to be able to distinguish between :
 - total Hg and monomethyl Hg
 - total As and arsenite, arsenate, monomethylarsonic acid, dimethylarsinic acid and arsenobetaine



Total As 0.2-76 mg/kg ww



mg/kg ww





Arsenic in seafish :

Total concentrations are much higher than the legal limits (varying from 0.1 mg.kg⁻¹ in Venezuela to 10 mg.kg⁻¹ in Hong Kong)
BUT toxic fractions constitutes only 1-2 % of the total concentrations
THUS speciation information is needed to investigate possible harm of seafood ingestion to human health

- Normalization was applied to divide species in classes of harmless to harmful
- benthic fish (flatfish) generally contain more As than pelagic fish
- Differences in As-concentrations according to sampling area could be found for some species
- Unlike for Hg, no evidence of biomagnification could be found for As



BIOGEOCHEMISTRY OF NUTRIENTS, TRACE METALS AND ORGANIC MICROPOLLUTANTS IN THE NORTH SEA.

SUBPROJECT: BIOGEOCHEMICAL BEHAVIOUR OF PARTICULATE TRACE ELEMENTS IN THE NORTH SEA

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Area of Investigation







Distribution of particulate elements

Spatial and temporal variations of the composition of the suspended matter in terms of major (Na, K, Ca, Mg, Fe, Si, Al, Corg, Cinorg) and minor elements (Mn, Cr, Ni, Co, Cu, Zn, Cd, Pb) in the Southern North Sea.

Major elements in relation with aluminium



Trace metals in relation with aluminium



Distribution of the log K_D



Affinity for the particulate phase: Pb > Zn > Ni > Cu > Cd

Acidimetric titration

Investigation of the mobility and thus the potential bio-availability of the particulate trace metals by performing batch titration experiments of suspended matter by a strong acid (1N HCI).



Major elements



Trace metals



Associated elements



Transfer of trace elements between the dissolved and particulate phases

Study of the mechanism and rate of uptake of trace metals by suspended matter using radionuclides (⁵⁴Mn, ⁶³Ni, ⁶⁵Zn, ¹⁰⁹Cd) during incubation experiments of natural samples.

¹⁴C incorporation experiments (July 1999)



Rate: 1.0 µg C I⁻¹ h⁻¹





Photosynthetic parameters

" = 0.023 : q C
$$I^{-1} h^{-1}$$
 (: E $m^{-2} s^{-1}$)







Rate: 22 µg C I⁻¹ h⁻¹

Use of specific conditions

CONDITIONS	PROCESSES
Light	Photosynthesis
	Photoreaction
Dark	Abiotic
	Bacterial
short term incub.	Isotopic exchange

|--|

INHIBITOR	INHIBITION
DCMU	Photosynthesis
Antibiotics	Bacterial activity
Sodium Azide	Biotic process

Radionuclides incorporation experiments - Trace Metals

Influence of light intensity



- No influence of light intensity except a slight photoinhibition of Mn transfer

Influence of biological activity - Inhibition by sodium azide



- Strong influence of biological activity on Mn uptake
- Strong abiotic uptake of Zn

SPECIFIC INCORPORATION





K= 62 10⁻³ h⁻¹



CONCLUSIONS

• In the coastal area under the influence of the Scheldt plume, the concentrations of particulate trace metals may be higher than that of the dissolved species, partly due to high values of the partition coefficient Kd. Most of the trace metals are associated with detrital alumino-silicates and Fe/Mn oxy-hydroxides of continental origin.

• Titration experiments with HCI developed during our study allowed a better evaluation of the mobility of the various particulate elements. In particular, it has been shown that the dissolution of most trace elements begins as soon as the pH drops below 7. However, in the case of Pb and Cr, the dissolution starts only at pH below 5.

• The use of radioactive trace elements allowed the assessment of biotic and abiotic fast reactions transferring the metals between the dissolved and particulate phases. As a consequence, the residence times of dissolved trace elements in surface waters are short (of the order of a few days). The particulate phase may act as a buffer to maintain the stock of dissolved elements constant.

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